

light is made of Colors

some light is invisible: Ultraviolet or Infrared

adding colored light

R

subtracting colors

Atoms or molecules Excited State absorb energy and reach an excited state. Then they release energy as light, electricity, or electricity, or heat as they return to the ground state. Ground State

heat



Silicon solar cells

are efficient but

expensive.

Dye sensitized solar cells and organic solar cells use chemicals to trap light.

if a substance absorbs light and then emits light at a different color, that is called Luminescence



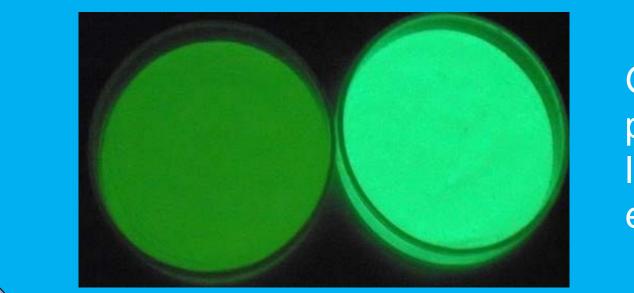
Quantum dots in solution fluoresce under UV light at different colors than they absorb. They can be combined with other materials to absorb light.

after chemicals absorb energy they can Emit Light



We can design chemicals that absorb and emit light at exactly the wavelengths we need. The color of fireworks comes from different elements.

460 470 480 490 500 510 520 530 540 550 560 570 580 590 600 610 620 630 640 650 660 670 680 690 700



Glow in the dark paint phosphoresces for a long time after it is excited by light.

if a material is very hot and emits light, that is called Incandescence





The color of incandescent light varies according to the temperature. Red is cool, white is hot.



Ultraviolet light from a UV A radiant heater uses near infrared, a remote lamp, or from the sun, is control uses far infrared powerful. radiation.



When red green and blue light are mixed you get white. This is how a TV makes white light.

Yellow, magenta and cyan colored inks or filters remove light to make any color including black. Where there is no ink the white paper shows through.

Μ

visible light is part of the Electromagnetic Spectrum

Type of EM Radiation	g Frequency (v)	ncreasin	← In							
^{z)} Gamma Rays	10^4 10^2 10^0 v (Hz)	10 ⁶	108	10 ¹⁰	10^{14} 10^{12}	10 ¹⁶	10 ¹⁸	10 ²⁰	10 ²²	10 ²⁴
X rays	Long radio waves	AM io waves		Microwave	IR	UV	X rays		γ rays	
ⁿ⁾ Ultraviolet	10^4 10^6 10^8 λ (m)	10 ²	10^{0}	10-2	10 ⁻⁶ 10 ⁻⁴	10 ⁻⁸	10^{-10}	10 ⁻¹²	10^{-14}	10^{-16}

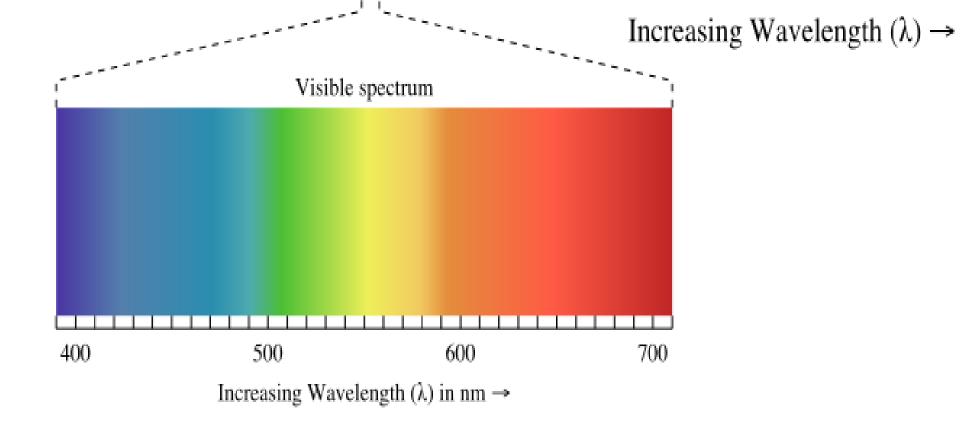
What it does- or is used for Most powerful radiation, causes cancer Passes through skinused to photograph the insides of things

if a chemical reaction **OCCUIS:**

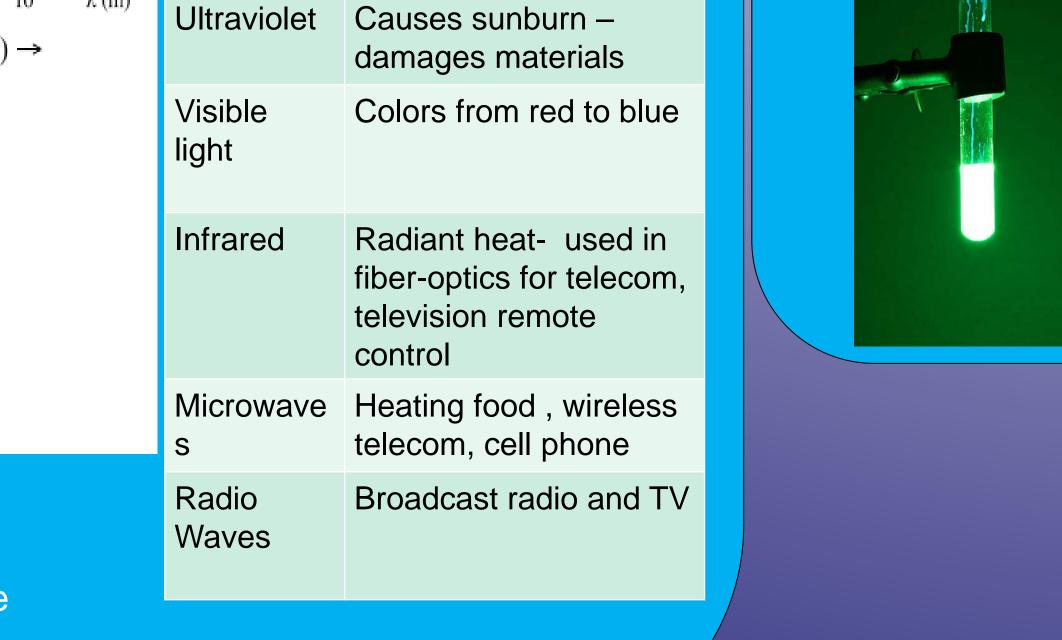
Chemiluminescence

if electricity excites the material: Electroluminescence





Electromagnetic radiation is described by wavelength or frequency. Certain bands have names such as visible, UV or infrared. The shorter the wavelength, the more energetic the photons.







White LEDs are used for energy efficient lighting.

An organic light emitting diode

(OLED) uses organic chemicals.

UNIVERSITY of WASHINGTON

A liquid crystal display has red, green and blue filters over a backlight.

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